

# Charge Example Problems

Monday, March 17, 2014 11:15 AM

- How many electrons are in 3.5 micro Coulombs?

$$3.5 \mu\text{C} = 3.5 \times 10^{-6} \text{C}$$

$$3.5 \times 10^{-6} \text{C} \left| \frac{1 e}{1.6 \times 10^{-19} \text{C}} \right| = 2.19 \times 10^{13} e$$

$$\begin{aligned} 1 \text{ cm} &= 10^{-2} \text{ m} \\ 1 \text{ mm} &= 10^{-3} \text{ m} \\ 1 \mu\text{C} &= 10^{-6} \text{ C} \end{aligned}$$

A lightning bolt has a charge of 1 to 10 C. How many electrons is this?

$$1 \text{ C} \left| \frac{1 e}{1.6 \times 10^{-19} \text{ C}} \right| = 6.25 \times 10^{18} e$$

$$10 \text{ C} \left| \frac{1 e}{1.6 \times 10^{-19} \text{ C}} \right| = 6.25 \times 10^{19} e$$

A particle has a charge of +6 micro Coulombs. How many protons is this?  
What would happen if the charge is -6 micro Coulombs?

$$+6 \mu\text{C} = 6 \times 10^{-6} \text{C}$$

$$6 \times 10^{-6} \text{C} \left| \frac{1 p}{1.6 \times 10^{-19} \text{C}} \right| = 3.75 \times 10^{13} \text{ protons}$$

“ “ electrons

After charge an object it now has 9.7 Me. What is its charge in Coulombs?

$$9.7 \text{ Me} = 9.7 \times 10^6 e$$

!M 10<sup>6</sup>

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$$9.7 \text{ Me} = 9.7 \times 10^6 \text{ e}$$

1M  $10^6$   
1km  $10^3$   
1m

$$9.7 \times 10^6 \text{ e} \left| \frac{1.6 \times 10^{-19} \text{ C}}{1 \text{ e}} \right| = 1.55 \times 10^{-12} \text{ C}$$

An object has a charge of -12 nC. How many electrons is this?

$$1 \text{ nC} = 1 \times 10^{-9} \text{ C}$$

$$-12 \times 10^{-9} \text{ C} \left| \frac{1 \text{ e}}{1.6 \times 10^{-19} \text{ C}} \right| = 7.5 \times 10^{10} \text{ electrons}$$

↑  
neg = electrons